



Lesson: Density

Name:

Teacher:

Date:

Worksheet 2 – Density Concepts Worksheet

Instructions: Choose the correct answer for each question.

1. Density is calculated as:
 - a) $\text{Mass} \times \text{Volume}$
 - b) $\text{Mass} \div \text{Volume}$
 - c) $\text{Volume} \div \text{Mass}$
 - d) $\text{Mass} + \text{Volume}$
2. Which of the following liquids is densest?
 - a) Water (1 g/mL)
 - b) Honey (1.4 g/mL)
 - c) Vegetable oil (0.92 g/mL)
 - d) Rubbing alcohol (0.79 g/mL)
3. Why does oil float on water?
 - a) Oil is more viscous
 - b) Oil is less dense than water
 - c) Oil is polar
 - d) Water evaporates faster
4. In the Density Rainbow Experiment, what property causes liquids to layer?
 - a) Color b) Temperature c) Density d) Volume
5. Which statement is correct about density?
 - a) All liquids mix if poured together
 - b) Denser liquids sink below less dense liquids
 - c) Solids always float in liquids
 - d) Gases have the same density as liquids
6. A grape floats above water but below oil. This indicates:
 - a) Grape density < oil
 - b) Grape density > oil and < water



- c) Grape density > water
 - d) Grape density = water
7. Which factor does **not** affect density?
- a) Mass of the substance
 - b) Volume of the substance
 - c) Color of the substance
 - d) Packing of molecules
8. Rubbing alcohol sits on top in the density experiment because:
- a) It is colored
 - b) It is less dense than the other liquids
 - c) It reacts with water
 - d) It is heavier
9. Which of these is a correct SI unit for density?
- a) g/mL b) mL/g c) L × kg d) g + cm³
10. Why is it important to understand density in real life?
- a) To cook food faster
 - b) To predict floating and sinking, pollution spread, and material design
 - c) To measure temperature
 - d) To color liquids



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Answer Key – Worksheet 2

1. b) $\text{Mass} \div \text{Volume}$
2. b) Honey (1.4 g/mL)
3. b) Oil is less dense than water
4. c) Density
5. b) Denser liquids sink below less dense liquids
6. b) Grape density $>$ oil and $<$ water
7. c) Color of the substance
8. b) It is less dense than the other liquids
9. a) g/mL
10. b) To predict floating and sinking, pollution spread, and material design